## Comment on vibrational predissociation of polyatomic van der Waals complexes

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Levy has utilized the energy gap law (EGL) for vibrational predissociation (VP) of triatomic van der Waals molecules (VDWMs)<sup>2</sup> to discuss the short lifetime<sup>3</sup> of the interesting polyatomic complex HeNO<sub>2</sub>. He points out that nonadiabatic interstate scrambling 4,5 between the excited <sup>2</sup>B<sub>2</sub> state and the ground <sup>2</sup>A<sub>1</sub> state of NO2 results in a level structure which is characterized by a relatively high density of vibronic levels (~0.05 cm), whereupon the effective energy gap between adjacent vibronic levels is considerably reduced. We would like to point out that such a qualitative application of the EGL for the VP of a complex consisting of a rare-gas atom X bound to a polyatomic molecule M, which is characterized by effective interstate mixing, requires the incorporation of anharmonicity effects. Consider a VDWM M-X which is characterized by the Hamiltonian for the internal motion  $H = H_0 + V$  with  $H_0 = H_M$  $+H_{X,M}$ , where  $H_M$  is the Hamiltonian for the conventional molecule M,  $H_{X,M}$  denotes the (zero-order) X-M interaction and where X moves relative to the center of mass (c.m.) of M, which is frozen at its equilibrium nuclear configuration, and V corresponds to the residual interaction. The zero-order diabatic states of  $H_0$  are  $\mathbf{X}^{\mathbf{M}}\phi_I(R_{\mathbf{X},\mathbf{M}})$  and  $\mathbf{X}^{\mathbf{M}}\phi_{\epsilon}(R_{\mathbf{X},\mathbf{M}})$ , where  $\mathbf{X}^{\mathbf{M}}$  is the nuclear wavefunction of M, while  $\phi_l$  and  $\phi_\epsilon$  correspond to bound and continuum states of the VDW bond, respectively. For the hypothetical case of a harmonic molecule  $\mathbf{X}_{\{v_j\}}^{\mathbf{M}} = \Pi_{j \chi_{v_j}}(q_j)$ , where  $\chi_{v_j}(q_j)$  is the nuclear wavefunction for the jth normal mode, specified in terms of the normal coordinate  $q_i$  and the vibrational state  $v_i$ . On the basis of our numerical calculations for X-BC<sup>2</sup> as well as for BC-BC<sup>6</sup> we assert that the dominant contribution to the perturbation Hamiltonian involves linear terms in the internal displacements  $X_{\delta}$ ,  $V(X_{\beta}, R_{c.m.}) = \sum_{\beta} X_{\beta} f_{\beta}(R_{X, M})$  so that for the harmonic molecule  $V = \sum_{i} q_{i} g_{\beta}(R_{X,M})$ , where  $f_{\beta}$  and  $g_{\beta}$  are some functions of the separation  $R_{X,M}$  between X and the c.m. of M. It is evident that the discrete-continuum coupling for the harmonic molecule is determined by the selection rules  $\Delta v_i = \pm 1$  for a single mode, while  $\Delta v_b = 0$  for

 $k \neq j$ . This simple selection rule leads to a propensity rule for the VP process which essentially involves one quantum jump within a normal mode of the harmonic molecule. Thus, the VDWM consisting of a harmonic conventional molecule exhibits the EGL separately for each normal mode of M. The Levy hypothesis can be realized provided that anharmonicity effects are incorporated. One possible representation of the nuclear wavefunction of  $\mathbf{X}^{\mathbf{M}}$  involves a harmonic basis

$$\mathbf{X}_{\alpha}^{\mathrm{M}} = \sum_{\nu_{1}} \sum_{\nu_{2}} \cdots C_{\nu_{1}\nu_{2}}^{(\alpha)} \cdots \chi_{\nu_{1}}(q_{1}) \chi_{\nu_{2}}(q_{2}) \ldots$$

These wavelengths, together with the perturbation  $[V(X_{\beta},R_{X-M})]$ , will result, obviously, in quantum jumps between close-lying nuclear states  $X_{\alpha}^{M}\phi_{l}$  and  $X_{\alpha}^{M}\phi_{\epsilon}$ , the coupling being determined by the expansion coefficients  $C_{v_{1}v_{2}}^{(\alpha)}$  and  $C_{v_{1}v_{2}}^{(\alpha')}$ . Obviously, this expansion for  $X_{\alpha}^{M}$  is not unique, and alternative representation, e.g., in terms of the zero-order bond modes, seems promising. Thus, anharmonicity effects are crucial for the reduction of the effective energy gap in the VP of polyatomic VDWMs. These anharmonicity effects have interesting implications for the study of intramolecular vibrational energy exchange in a normal M molecule interrogated by monitoring the VP of the X-M complex.

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