

The result Q_{MM}^* obtained by previous investigators is the sum of Q_1 and the first term of Q_2 . (Note that $Q_{MM}=Q_1+Q_2$ reduces to Q_{MM}^* in the classical limit $\hbar \rightarrow 0$.) When μ , C , and/or v are small, the higher-order terms in Eq. (3) become important. For the purpose of illustrating the importance of these terms we consider $s=6$. For this case, $Q_{MM}=7.596(C/\hbar v)^{\frac{1}{2}}(1+25.480X+2.114X^2+10.031X^3+\dots)$, where $X=(\hbar^2/\mu C)(C/\hbar v)^{\frac{1}{2}}$ and $7.596(C/\hbar v)^{\frac{1}{2}}=Q_{MM}^*$. The calculated values for $^6\text{Li-Xe}$ and $^7\text{Li-Kr}$ are plotted in Fig. 1 along with the result of the partial wave analysis, Q_{PW} , of Rothe, Rol, and Bernstein² as a function of v . In these two systems $X=1.151 \times 10^2/v^{\frac{1}{2}}$ and $1.138 \times 10^2/v^{\frac{1}{2}}$, respectively; therefore, the correction terms in Q_{MM} resulting from the inclusion of the terms in $\eta(l, E)$ with $n \geq 2$ play an important role in controlling the over-all value of the cross sections. For example, when $v=10^5$ and 4×10^5 cm/sec, $Q_{MM}/Q_{MM}^*=1.293$ and 1.096 , respectively, in $^6\text{Li-Xe}$ and 1.290 and 1.095 in $^7\text{Li-Kr}$. The ratio is insensitively dependent on the nature of the collision partners of the present type. In Fig. 1 it may be interesting to note that in the "intermediate" velocity range Q_{MM} and Q_{MM}^* vary as if they are the upper and lower bounds of the undulatory results of Q_{PW} . The deviation of Q_{PW} from Q_{MM} decreases with increasing v : $Q_{PW} \simeq Q_{MM}$ for $v > 3 \times 10^6$ cm/sec in $^6\text{Li-Xe}$ and for $v > 2 \times 10^6$ cm/sec in $^7\text{Li-Kr}$ in the velocity range considered here.

Although it is important to consider the quantum interference^{6,7} at low v , the MM method can still give us valuable information about the magnitude of Q and its velocity dependence, if we include the higher-order terms of $\eta(l, E)$. The results obtained in the past²⁻⁴ in terms of Q_{MM}^* must be corrected by the additional important terms shown in Eq. (3).

The Landau-Lifshitz (LL) method⁸ also considers the leading term of $\eta(l, E)$ only, i.e., Q_{LL}^* . Although Bernstein and Kramer⁹ showed that Q_{LL}^* is superior to Q_{MM}^* , it is not possible to compare these two methods unless we explicitly include the higher-order terms. While we are able to calculate Q_{MM} including all important higher-order terms, Q_{LL} is very difficult to calculate due to the mathematical complexity involved in the integration of

$$\int_0^\infty l \sin^2 \eta(l, E) dl,$$

with $\eta(l, E)$ given by Eq. (2). Still another approximate formula is that of Schiff¹⁰ but it becomes identical to Q_{LL} for a spherically symmetric potential, when we introduce the small angle approximation.

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Conjecture on Resonant Transfer of Vibrational Energy and Radiationless Transitions in the Solid Phase: The Lifetime of Triplet Anthracene

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THE currently accepted description of radiationless processes in the solid state at low temperatures is as follows^{1,2}: a molecule in an excited vibronic state loses vibrational energy, thereby dropping to the zeroth vibrational level of the excited electronic state. From this state there is a transition, by tunneling between the potential-energy surfaces of the excited and ground electronic states, to a vibrationally excited state of the ground electronic state. Finally, the excess vibrational energy of the ground electronic state is removed by the surrounding medium. It is usually also assumed that the loss of vibrational energy is very rapid, so that the rate of radiationless transition is determined by the tunneling step which, in turn, is related to the product β_n of the matrix element of the perturbation which mixes the two states, β_{el} , and the Franck-Condon factor $F^{\frac{1}{2}}$. Except under unusual circumstances, involving strong solute-solvent interaction, it is a consequence of this last assumption that the rate of radiationless transition will be independent of the nature of the surrounding (dense) medium.

However, recent studies of laser excitation of crystalline anthracene^{3,4} and naphthalene⁵ show that the lifetime of the lowest triplet state is much shorter in the crystal than in an EPA glass. (Anthracene $\tau_{\text{xtal}}=0.01$ sec,³ $\tau_{\text{glass}}=0.1$ sec,⁴ naphthalene $\tau_{\text{xtal}}=0.02$ sec,⁵ $\tau_{\text{glass}}=2$ sec.) Assuming these data to be correct, it is clear that a pronounced medium effect exists and that the existence of this effect requires re-examination of the accepted theory.

In this note, we conjecture that the origin of the medium effect on the triplet lifetime arises from the difficulty of removing vibrational energy from the "hot" molecule in a glass. That is, we examine the possibility that the rate of transfer of vibrational energy is not rapid relative to the rate of intersystem crossing. It is assumed herein that in the pure crystal intermolecular vibrations are coupled via resonance interactions in-

volving vibrational transition multipole moments. Because the vibrational exciton bandwidth is expected to be of the order of 1 cm^{-1} , an efficient channel for vibrational relaxation in the crystal is available. Anharmonicity effects in large polyatomic aromatic molecules, where many atoms are involved in the normal vibrations, are usually extremely small⁶ and will not lead to a mismatch of the energy levels.

Consider the case of crystalline anthracene. Provided that the vibrational exciton bandwidth I is large relative to β_n , the rate of transfer of vibrational energy may be shown to be^{2,7} $\mathcal{R} = \beta_{e1} F / \hbar I$. The necessary Franck-Condon factors may be computed after the vibrational overlaps for skeletal stretching, CH stretching, and carbon-atom and CH out-of-plane bending motions are evaluated. Using $\beta_{e1} \approx 0.01\text{ cm}^{-1}$, and computing $I \approx 0.2\text{ cm}^{-1}$ for the quadrupole coupled vibrational exciton band, we find⁸

$$\mathcal{R} \approx 10^2\text{ sec}^{-1}$$

which leads to a predicted triplet lifetime of 10 msec, in fortuitous agreement with experiment. The quoted value of \mathcal{R} is uncertain to an order of magnitude (possibly two), but is still reliable enough, in our opinion, to show that the nature of the coupling of the hot molecule to the medium may play an important role in determining the rate of a radiationless transition.⁹

Putting aside quantitative details, what qualitative predictions can be made? First, the coupling of a vibrationally hot molecule to a glassy medium is smaller than the corresponding coupling in the pure crystal because, when the matrix is of different species from the hot molecule, resonant transfer of energy does not occur. Second, it is important that the vibrational energy be removed by free-running excitons because (a) a lattice deformation tends to decrease I greatly, (b) in the new quasistationary state of molecule plus deformed medium, β_{e1} may be altered by mixing in of new states, (c) when I is not much greater than β_n , the analysis leading to the rate equation cited is no longer valid with the result that a single vibronic state is not coupled to a band of host vibrational states.

In view of the preceding comments, it would be interesting to know if the lifetime of the triplet of 2,2-paracyclophane differs much between EPA glass and pure crystal. Our analysis leads to the prediction that it should not, since this compound forms an internal triplet excimer¹⁰ and the crystal structure is such that there ought to be a very small vibrational bandwidth. Indeed, we are led to suggest, on the same grounds, that the triplet of crystalline [2,2-] paracyclophane may be long lived relative to the triplet of crystalline benzene.

We conclude that the assumption that I is always sufficiently large that the rate-determining step in a radiationless transition is the tunneling between potential-energy surfaces is not, *a priori*, justified. Indeed, there seem to be cases where the experimental data require a different interpretation.

It is, of course, possible that β_{e1} is not independent

of the surrounding medium, and that the approximation of writing β_n as a product is not sufficiently accurate because of changes in coupling in the highly excited vibrational states. The magnitudes of these effects remain to be explored.

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⁶ In the case of the α_{1g} vibration of benzene, the anharmonicity factor is about 0.5 cm^{-1} [F. M. Garforth and C. K. Ingold, *J. Chem. Soc. (London)* **1948**, 417]. The anharmonicity factors for naphthalene and anthracene are expected to be smaller.

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⁸ The estimate for β_{e1} comes from Ref. 2. To compute the vibrational overlap integrals, the method of E. F. McCoy and I. G. Ross [*Australian J. Chem.* **15**, 573 (1962)] was used to estimate changes in the origins of the normal modes in the excited state. E. Hutchisson's formula was then used [*Phys. Rev.* **36**, 410 (1930)]. Normal mode frequency changes in the excited state were estimated from experimental data on analogous systems, and C. Manneback's formula used to compute the overlap integrals [*Physica* **17**, 1001 (1951)]. The estimated vibrational bandwidth agrees with the few data available for analogous systems [e.g., T. M. Nedungadi, *Proc. Indian Acad. Sci. Sec. A* **15**, 376 (1942)].

⁹ Pure crystals of benzene, naphthalene, and anthracene do not phosphoresce. Although this lack of phosphorescence is usually attributed to efficient triplet-triplet annihilation, examination of the data obtained in Ref. 3 shows that the biexciton annihilation process does not alter the effective lifetime of the triplet state for triplet-exciton concentrations smaller than 10^{13} cm^{-3} . We conclude that the decrease in the lifetime of the triplet state arises from an efficient monomolecular decay process, in agreement with the arguments we have advanced.

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Cross Section for Atom Exchange with Charge Transfer in a Molecular Collision*

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A MOLECULAR beam has been used to study the exchange reaction



from threshold (11.9 eV) to 60 eV. The absolute cross section was determined by total negative-ion collection, and a mass spectrometer was used to monitor the NO^+ ions.